

Equilibration des états de spins nucléaires dans l'espace : apports des études en laboratoire

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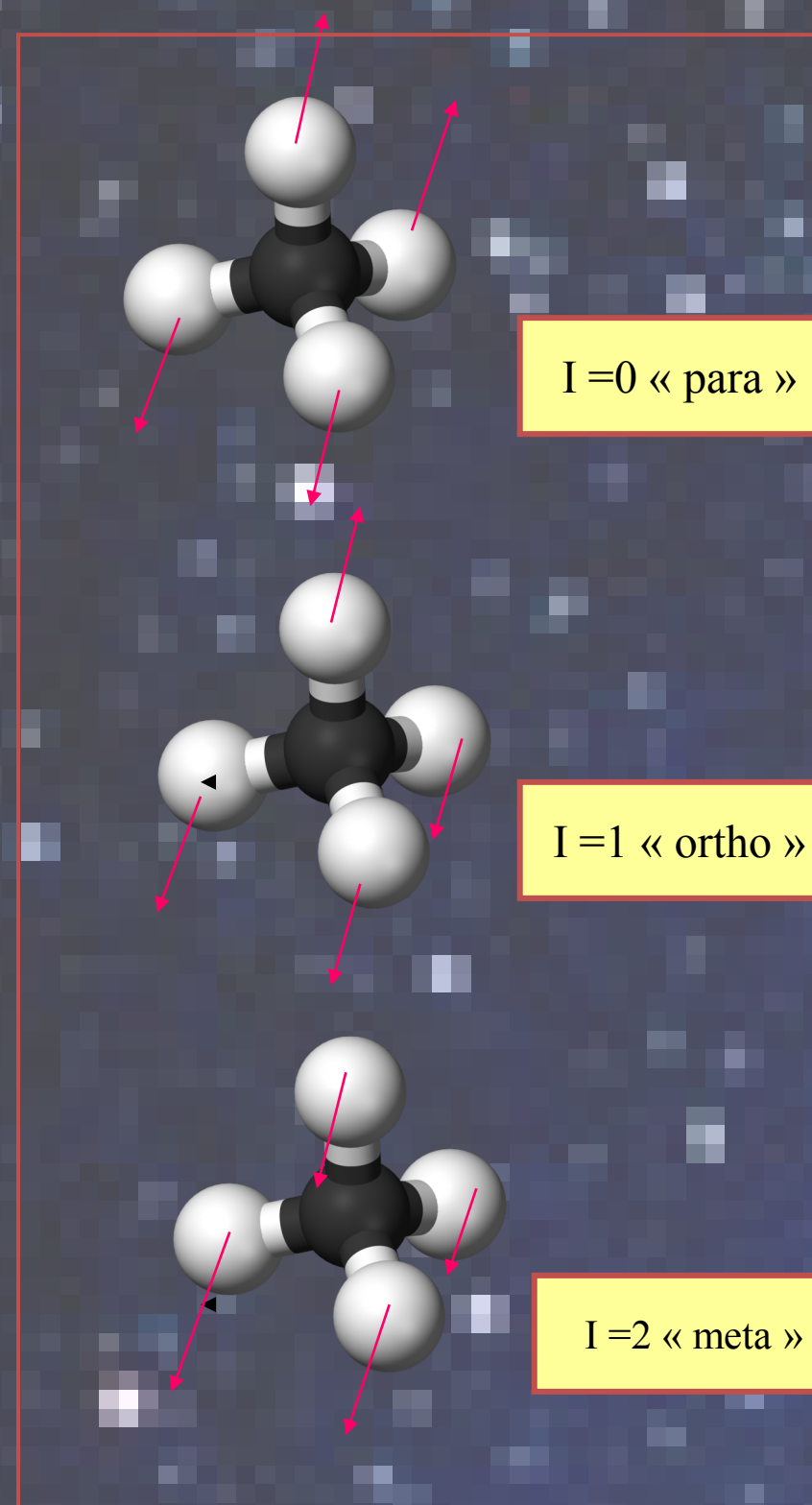
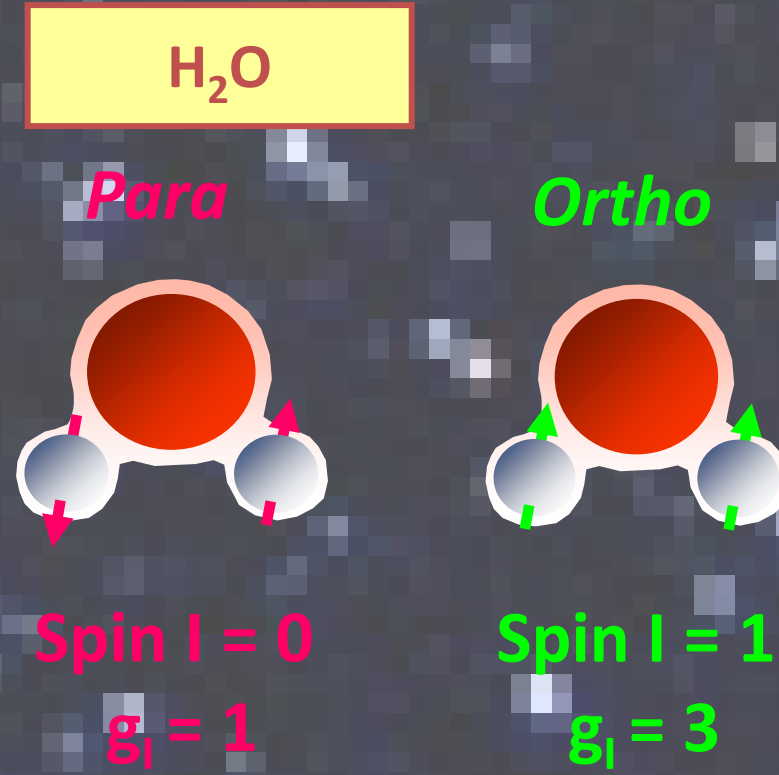
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GAS SPIN

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1. Spin Isomers

Due to the Pauli exclusion principle, hydrogenated molecules such as H₂O, having equivalent atoms exist in different nuclear spin configurations (also called "spin isomers") corresponding to different values of the total nuclear spin I.

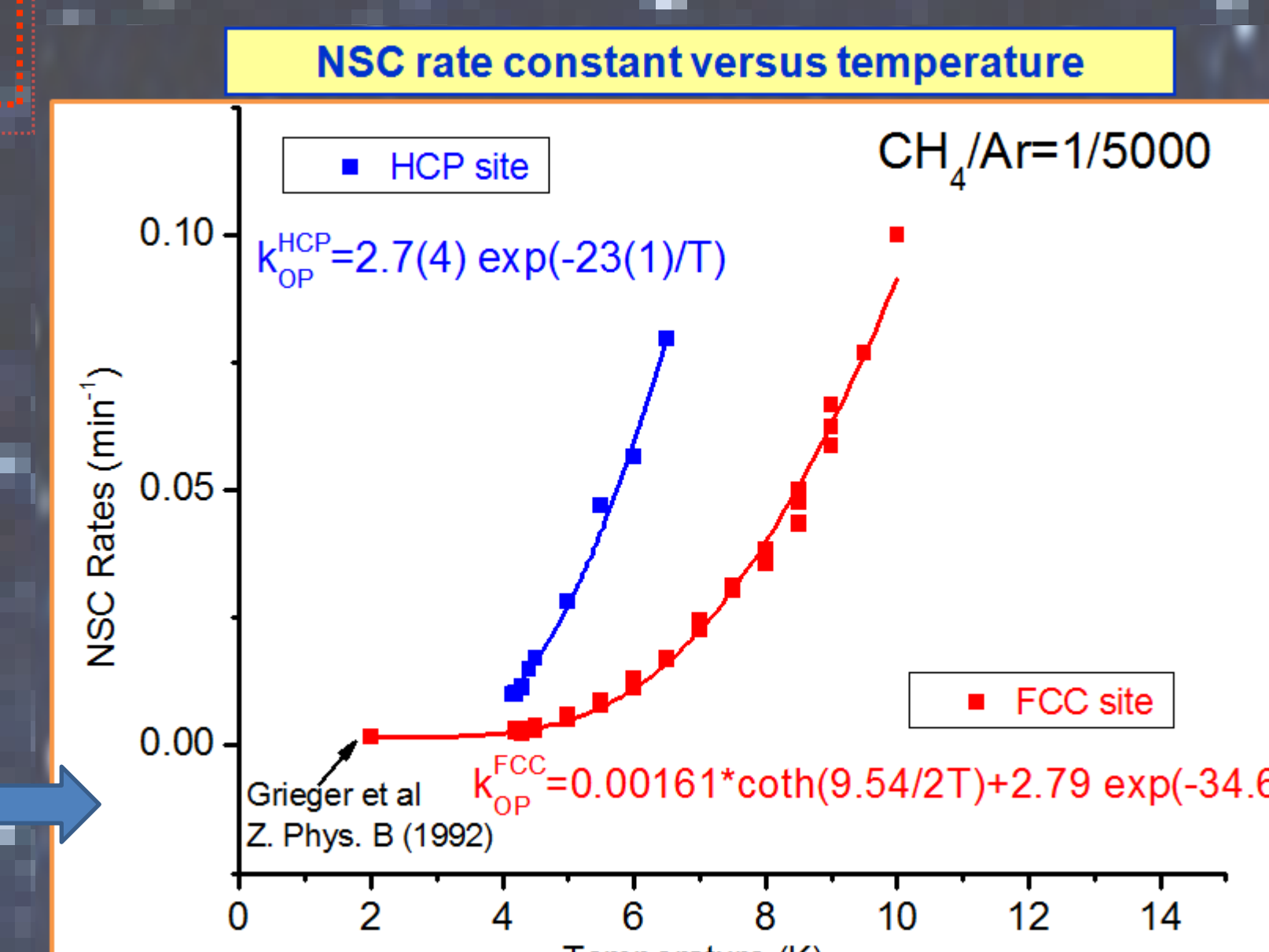
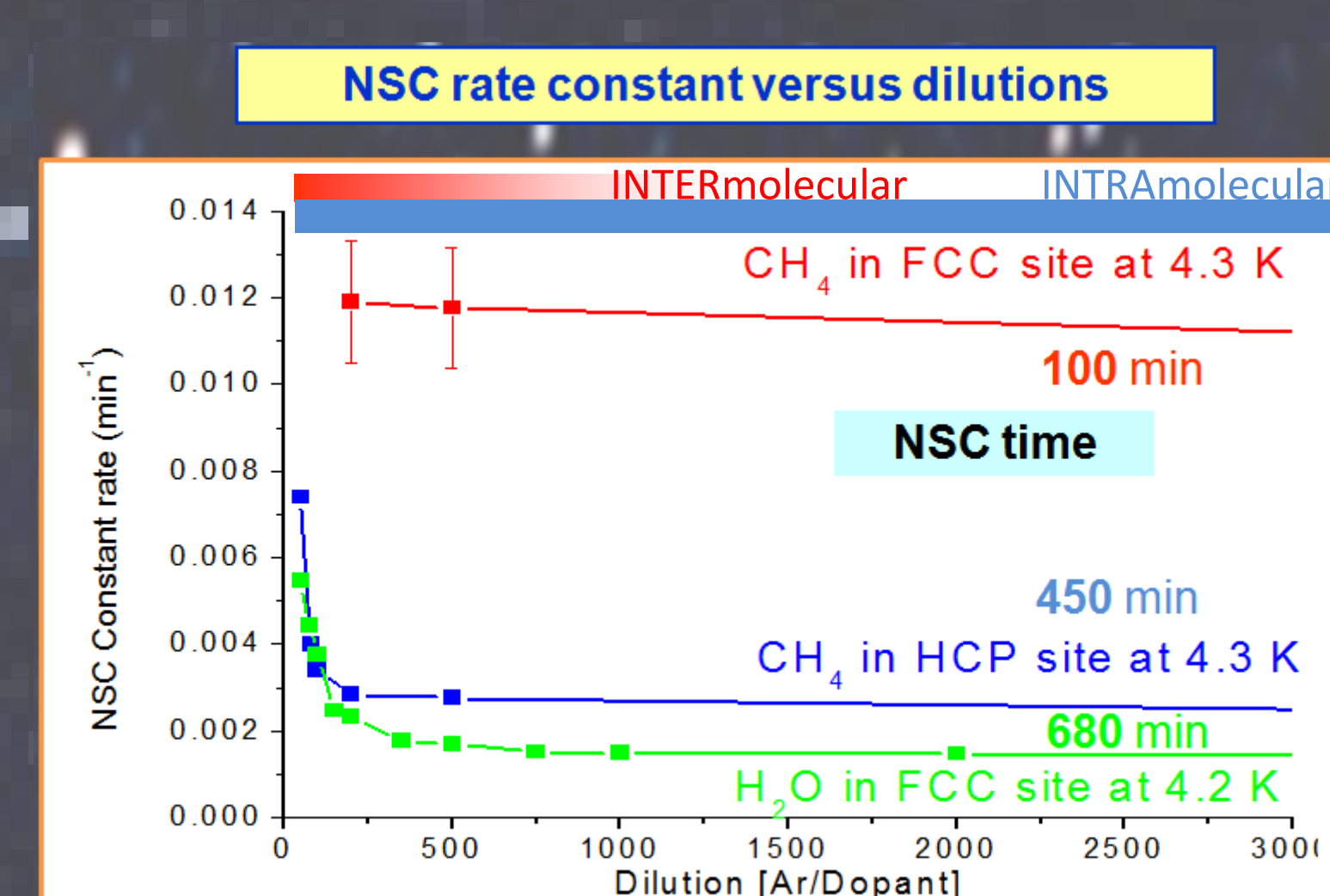
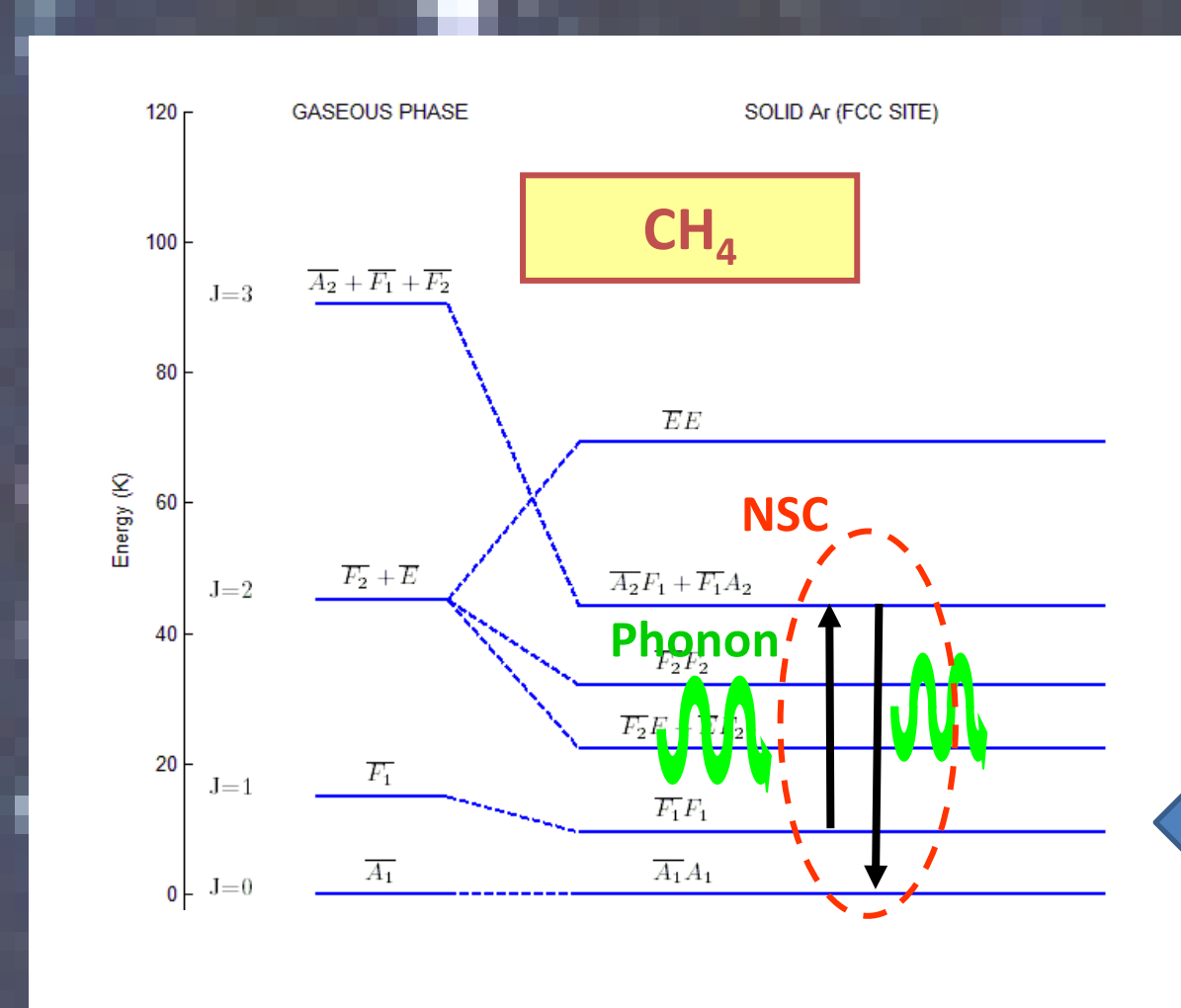
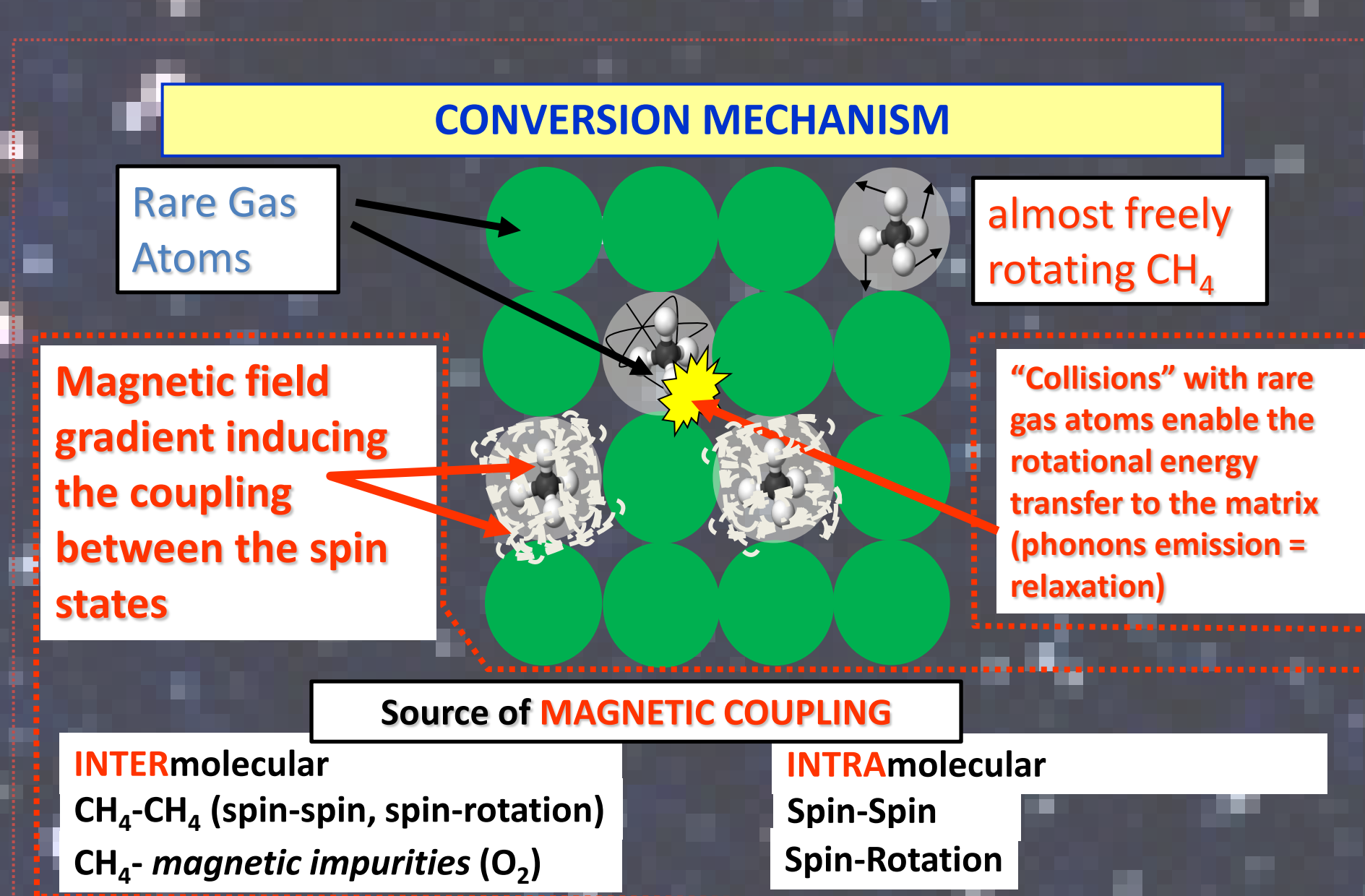


Due to the half-integer value of the spin of the proton, methane possesses 3 spin isomers with total spin I = 0, 1 or 2. Each spin configuration has specific symmetry characters (E, F, or A) of the symmetry operations of the group T.

| species | E | F | A |
|--------------------|---|---|---|
| Total nuclear spin | 0 | 1 | 2 |

| species | A ₁ | F ₁ | E+F ₂ |
|-----------------------------|----------------|----------------|------------------|
| Rotational Quantum Number J | 0 | 1 | 2 |

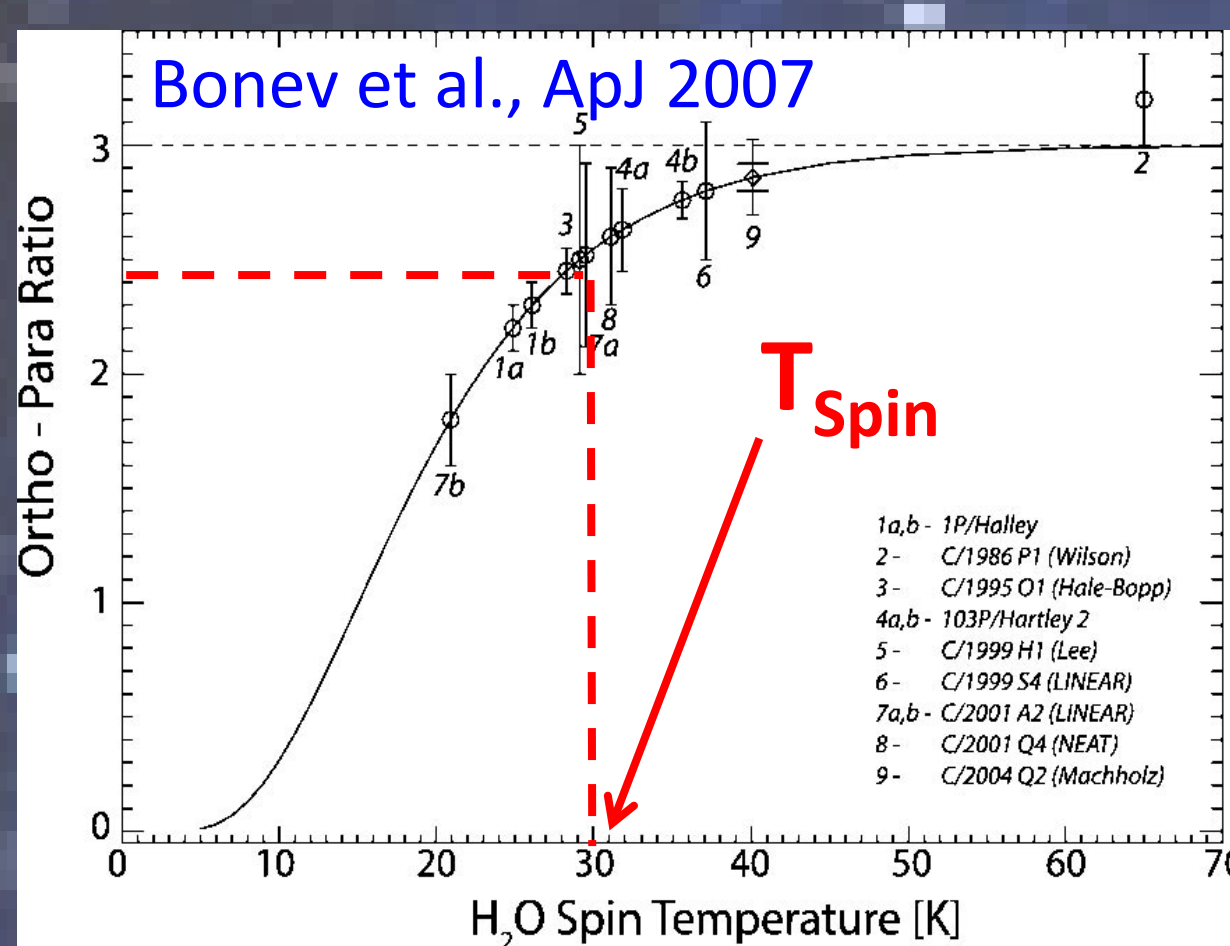
3. Nuclear Spin Conversion of Rotating Molecules in Solid State



NSC conversion is faster for methane than for water presumably due to the small energy difference (10 cm⁻¹) between the lowest rotational energy levels of *ortho* (J=1) and *meta* (J=0) species (to be compared to the 22 cm⁻¹ between *ortho* and *para* states of water). Strong differences are observed for CH₄ trapped in Face Centered Cubic and Hexagonal Close-Packed sites. They are also presumably due to the energy spacing between rotational states in the different sites. In the Ar matrix (FCC site), the rotational states of the molecules are distorted by electric interactions. The J=2 level is splitted in 3 sublevels with F₂ and E symmetry. Analysis of the temperature dependence using multiphonon Orbach model shows that in FCC site, NSC is enhanced by the degeneracy of the state J=3. In case of HCP site, the NSC is dominated by coupling of the J=1 state to a state lying 23/1.44=16 cm⁻¹ above. Temperature dependence of NSC in Ar matrix seems to follow a two phonons Orbach mechanism (Scott and Jeffries, *Phy. Rev.* 127, 32 (1962)). Sequentially, one phonon is absorbed, the molecule makes a rotational transition to an intermediate state before going to the final state with the emission of one phonon.

Nuclear Spin Conversion is fast in solid state and depends of the energetic structure of the molecule and the interactions with the environment. (Pardanaud *et al* J. Mol. Spec 2009, Abouaf-Marguin *et al* CPL 2009, Pardanaud PhD 2007, Letic PhD 2011)

2. Astrophysical context



OBSERVATIONS
nuclear spin statistic out of equilibrium

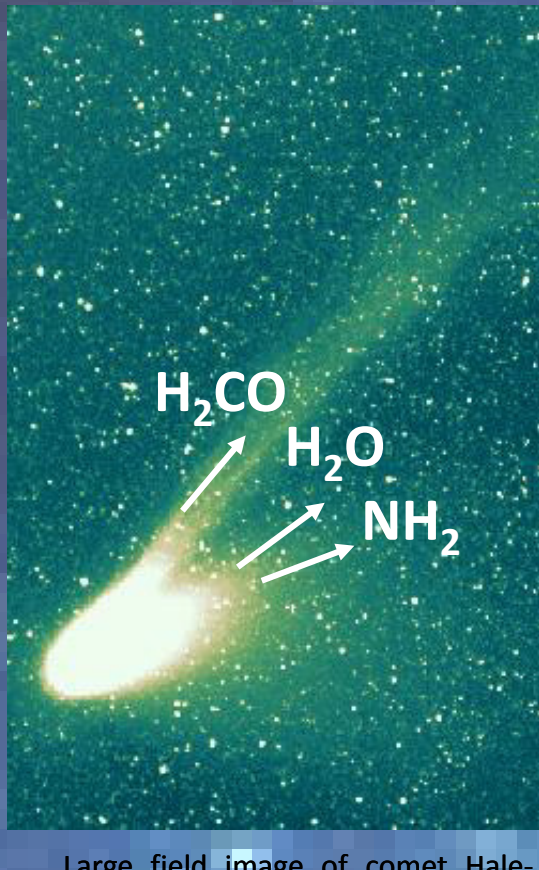
Physical origin of the spin temperature ?

$100 \text{ K} > T_{rot} > 50 \text{ K}$

$T_{spin} \sim 30 \text{ K}$

Memory of the spin states equilibrium at the formation date of grains or molecules ?

COMETARY ATMOSPHERES



H₂O

(C/1999 H1) : T_{spin} ≈ 30 K

(C/2001 A2) : T_{spin} = 23 ± 4 K

(Hale-Bopp) : T_{spin} = 28 ± 2 K

NH₃ (Hale-Bopp) : T_{spin} = 26 K

CH₄ (C/2001 Q4) : T_{spin} = 33 ± 3 K

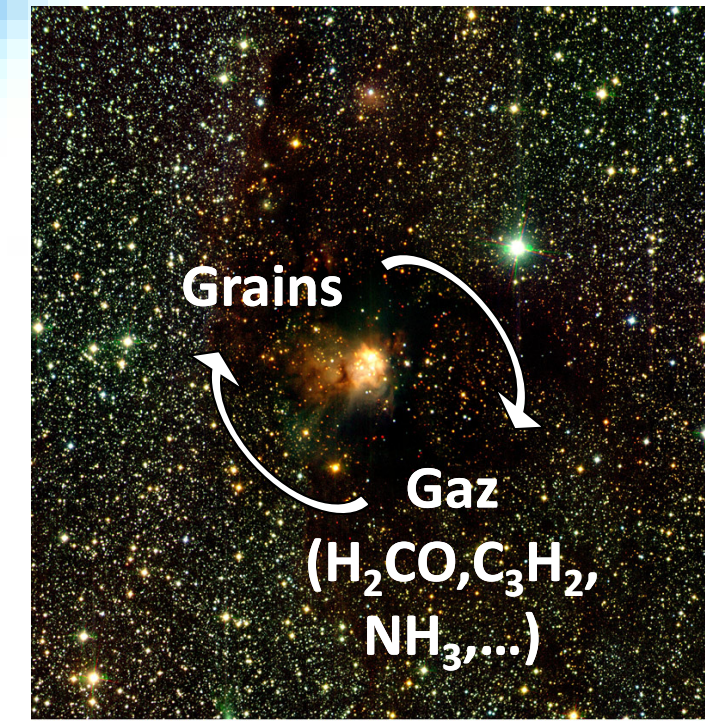
Solid-gas interface

Hydrogenated species

Hypothesis: NSC forbidden on the cometary core

« ... The small diversity of the nuclear spin temperatures and lack of clear correlation between T_{spin} and chemical composition in several comets are consistent with the hypothesis that T_{spin} reflects the temperatures in the presolar nebula. » Kawakita *et al.*, *Astrophysical Journal* 2006

INTERSTELLAR CLOUDS



NH₃ (L1157) : T_{spin} ~ 18 - 25 K Umemoto *et al.*, *Astrophys. J.* (1999)

H₂CO (L723,...) : T_K ~ 10 K Dickens *et al.*, *Astrophys. J.* (1999)

C₃H₂ (TMC-1) : OPR < 2 Morisawa *et al.*, *Astrophys. J.* (2006)

(correlated to chemical evolution)

H₂O (young star TW Hydrae) : T_{spin} ~ 14 K Hogerheijde *et al* *Science* (2011)

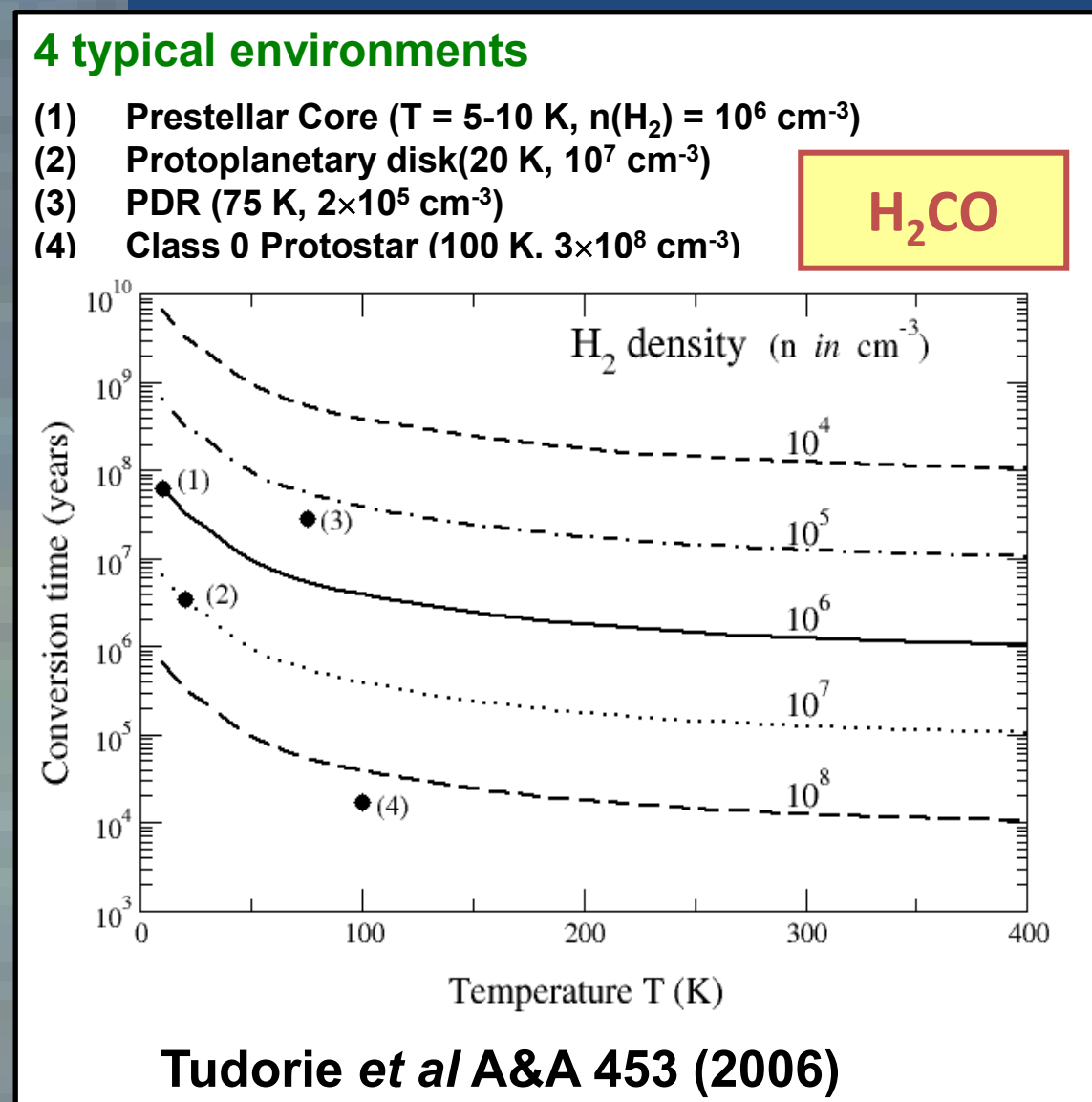
ADMITTED ASSUMPTION = NSC FORBIDDEN IN GAS PHASE BUT FAST ON GRAINS WHEN THE MOLECULE IS FORMED

What kind of information about the « history » of the molecule could we deduce from measurements of ratios between spin isomers? Shall we say that T_{spin}=T_{formation} ?

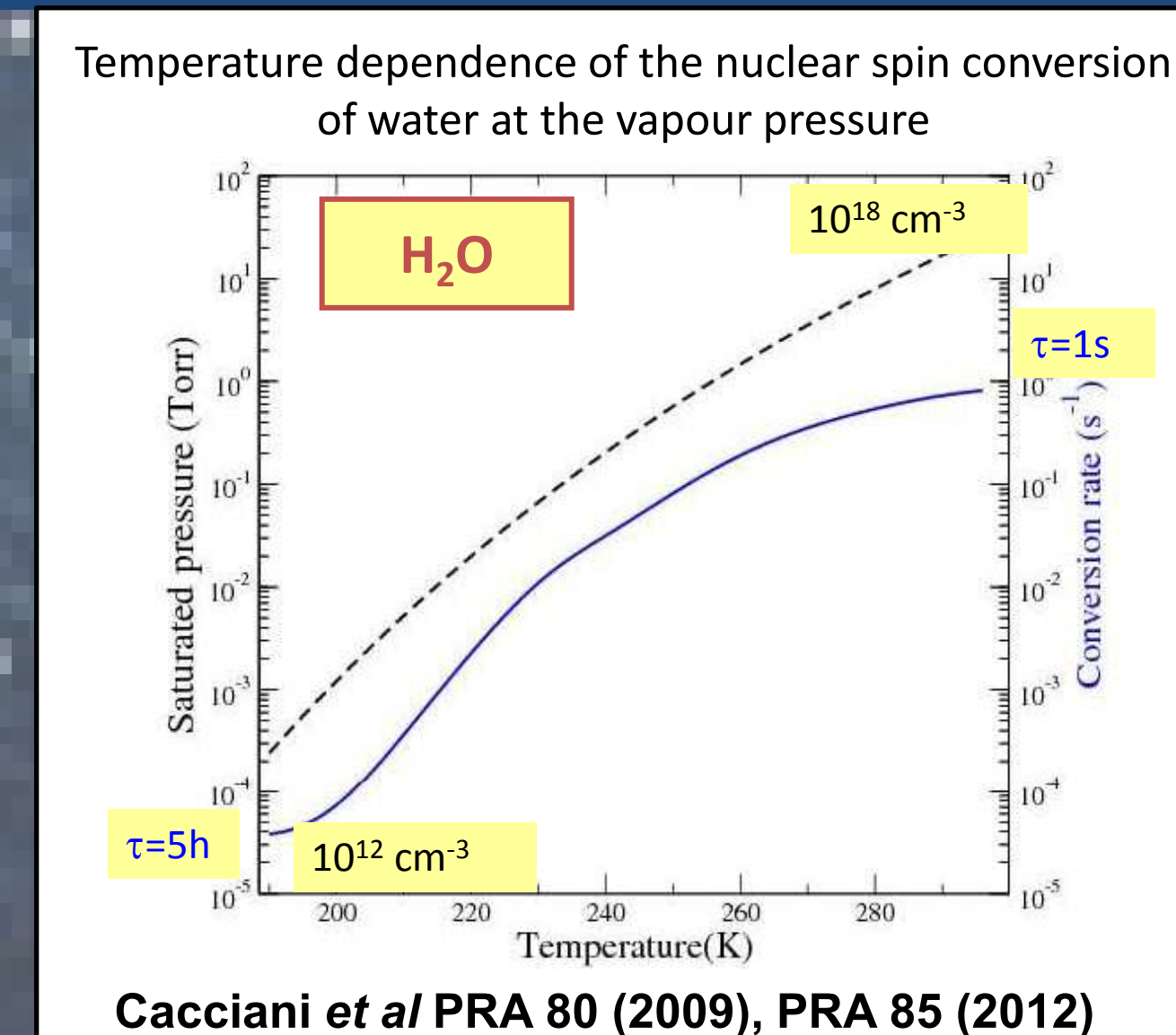
« ... The O/P ratio after a reaction reflects neither the temperature of reaction fields nor the excess energy of the reaction, but simply reflects the conservation of angular momentum. » Morisawa *et al.*, *Astrophys. J.* (2006)

« ... Radiative conversion between spin isomers is not allowed in the gas phase, preserving the OPR for long time scales. Gas-phase formation of water occurs through exothermic reactions leading to an OPR of 3. On grains, water forms and survives at low temperatures, and it is tempting to equate T_{spin} with the grain temperature. However, the energetics of water formation and ortho-to-para exchange on grains are poorly understood, and the water OPR may be changed by photodesorption. [...] Provided that spin temperatures reflect formation histories, the different T_{spin} inferred for the water ice in TW Hya (<13 K) and solar system comets (>20 K) indicates a similar mixing of volatiles throughout the entire solar nebula, blending water formed at >50 K and an OPR of 3 with water formed at 10 to 20 K and an OPR < 1 probed by our observations. In this case, the range of T_{spin} values of the cometary inventory reflects the stochastic nature of transport and mixing. » Hogerheijde *et al* *Science* (2011)

4. Nuclear Spin Conversion of Molecules in the Gas Phase



CALCULATIONS

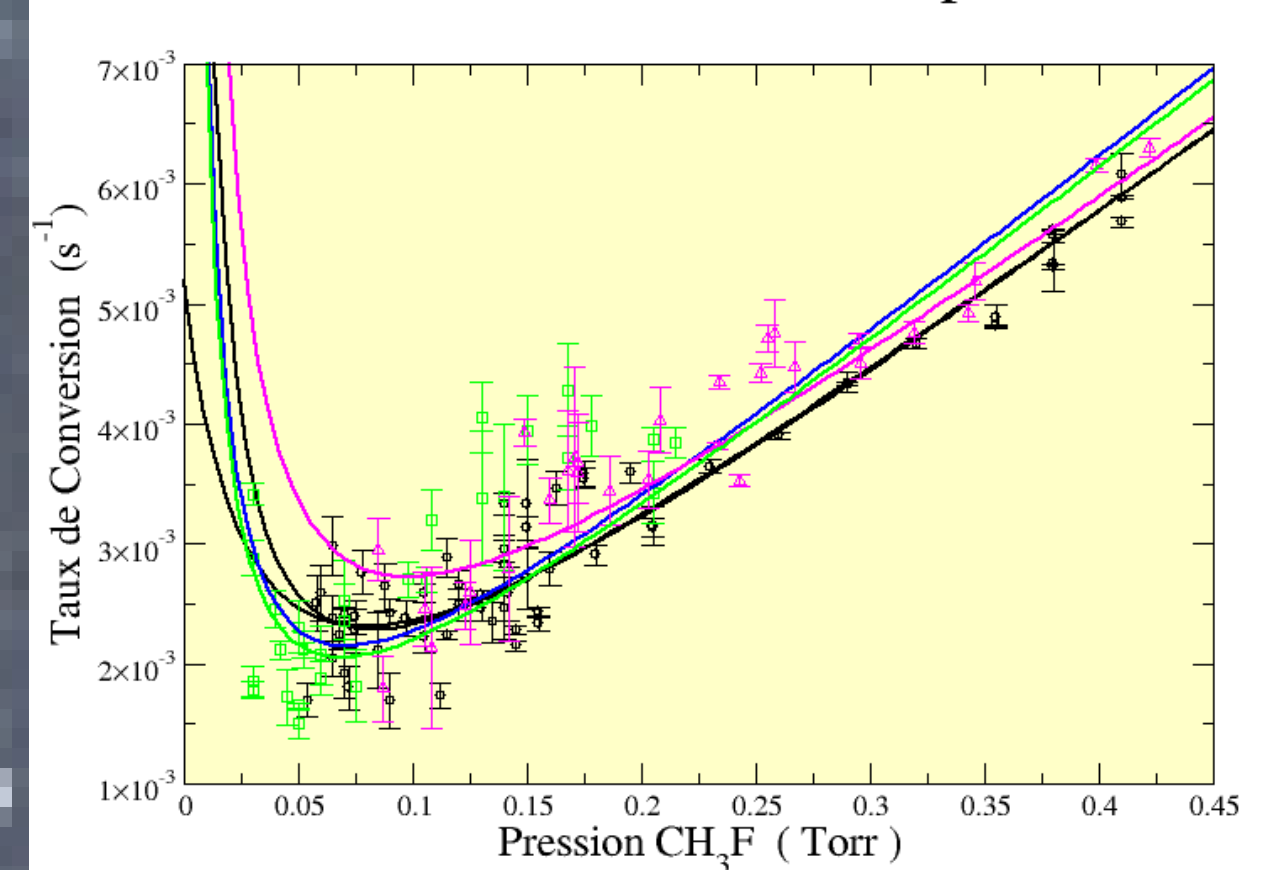


Calculations performed extrapolated to the case of water embedded in rare gas solid at 4.2 K are coherent with conversion time of about 10 h observed experimentally in dilute samples.

Magnetic intermolecular interactions that explain acceleration of the NSC in concentrated rare gas solid has to be implemented for the gas phase.

Calculations taking into account magnetic intramolecular interactions and collisional induced energy relaxation show that Nuclear Spin Conversion is very dependent on the the density and temperature of the gas phase.

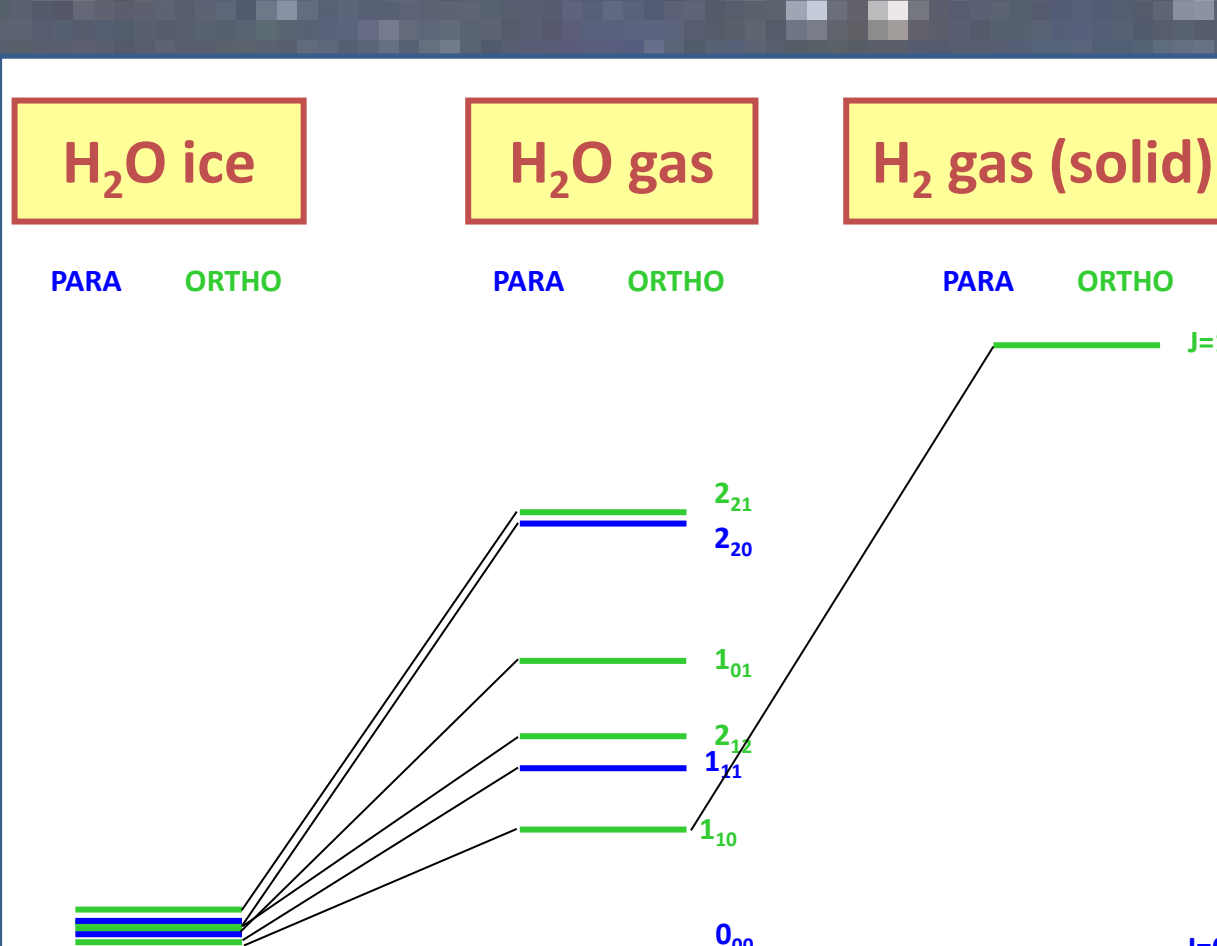
Taux de conversion à basse pression



EXPERIMENTAL

While the presence of electric fields in the vicinity of the molecule has been shown to accelerate nuclear spin conversion for CH₃F (Tudorie *et al* *Surf. Science* (2007), Cacciani *et al* *J. Mol. Spec.* 780 (2006)), we did not observe nuclear spin disequilibrium in the gas phase at low temperature (typ. 40-50 K) for CH₄ (Cermak *et al.*, *J. Mol Spec.* 2012).

5. Conclusions and Perspectives : GAS-SOLID interface



For H₂ trapped on Amorphous Solid Water (Chehrouri *et al* *PCCP* 13 (2011)), it was shown that without the presence of magnetic impurities the nuclear spin conversion is slow (τ > 4h) at low temperature. Thermal desorption does not induce important NSC in contrary of what it is observed for rotating molecules in rare gas solids.

The difference between the rotational energetic structures and the low ability of the icy surface to absorb the released energy may be crucial.

In the meanwhile, thermal desorption experiments for water on ASW (Hama *et al* *ApJL* 2011) have shown that high temperature OPR value is obtained since 150 K.

Disequilibria observed in space may be due to interactions in a mixture of solid and gas or due to the effects induced by UV or IR radiations. Development are in progress in LPMAA-Paris, PhLAM-Lille and PIIM-Marseille to answer this question.